# Purification and characterization of L-asparaginase extracted from local Iraqi green beans (*Phaseoulus vulgaris*)

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**Abstract**: The most prevalent metabolite for the storage and transport of nitrogen used in protein production is L-asparagine; in earlier studies, L-asparaginase enzyme derived from microorganisms was utilized to treat cancer cells; therefore, this study aimed to purify and describe this enzyme derived from Iraqi green bean seeds rather than microbial sources. The enzymes were partially purified from green beans by short steps, including centrifugation of crude enzyme, dialysis by Diethylaminoethyl Sepharose Column, and equilibrated using 20 mM Tris-HCl buffer, pH 8.0, and then applied to a sephacryl S-200. Coulometric methods measured the enzymatic activity at 450 nm, and the unit of activity was calculated by comparing it to a standard curve. Purification of L-asparaginase yielded a 5 percent yield, a 2.7 fold increase in activity, and a 43 unit/ml activity. The pH of asparaginase was optimal at 8.0. After a one-day incubation period, this enzyme became more stable at pH levels ranging from 7.5 to 9.5. This enzyme had the same optimal temperature and thermal stability at 40°C, but it was more stable at temperatures ranging from 20 to 40°C, allowing it to retain its maximal activity.

Key words: Asparaginase, phaseoulus vulgaris, leukemia.

# Introduction

In plants, L-asparagine is the most abundant metabolite for the storage and transport of nitrogen that is utilized in protein biosynthesis. Borek<sup>1</sup> pointed out that there are two known routes for L-asparagine metabolism for storage and transport of nitrogen utilized in protein synthesis. The first route, catalyzed by asparaginase, involves the hydrolysis of L-asparagine to release ammonia and L-aspartate. The second route involves the transamination of L-asparagine (in the presence of an oxo-acid) to form 2-oxosuccinamic acid and appears to be important in green leaves where it may play a role in photorespiration<sup>2</sup>. Lymphoblastic leukemia and non-Hodgkin's lymphoma were treated by L-asparaginase<sup>3</sup>. This enzyme catalyzes the conversion of L-asparagine into L-aspartate and ammonia, and this catalytic reaction is irreversible under physiological circumstances<sup>4</sup>. Asparaginase discourages protein production in cancer cells by forbidding them of the amino acid asparagine. First enzymatic hydrolysis of L-Asparagine to L-Aspartate and ammonia were reported by Lang<sup>5</sup>, who noticed L-Asparaginase activity in several beef tissues. L-Asparaginase has attracted much attention in both food industrial and pharmaceutical applications. In the food industry, L-Asparaginase was used to eliminate acrylamide from different processed foods using pretreatment of L-Asparaginase to degrading L-asparagine, the precursor of acrylamide, before baking<sup>6</sup>.

# Materials and methods

## **Plant-Based Materials**

The local Baghdad market provided mature green beans (*Phaseolus vulgaris*). Purification of Asparaginase.

# **Crude Extract**

Asparaginase extract in its most proper form was made by homogenizing 100g of green bean in 20 mM Tris-HCl buffer, pH 8.0 (including 1 mM PMSF, 12.5 mM -mercaptoethanol, 50 mM KCl, and 10% glycerol), spin at 10,000 rpm, and the crude extract was the name given to the supernatant<sup>7</sup>.

#### Diethylaminoethyl Sepharose Column

The dialyzed crude extract was put to a 16 \* 1.5 cm Diethylaminoethyl-Sepharose column that had been equilibrated before use with 20 mM Tris-HCl buffer, pH 8.0. Different concentrations of NaCl produced within a single buffer at a 1 mL/ minute flow rate were used to elute the enzyme, and 3 mL fractions were collected.

#### Sephacryl S- 200

Choose the elutions (fractions) from the previous step's peak that have the highest activity and apply them to a sephacryl S-200 (2.6\*83cm) column equilibrated with Tris-HCl at a flow rate of 0.5/minute, collecting 3.5 ml fractions.

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#### Asparaginase assay

The enzyme test reaction mixture contains 0.5 milliliters of pH 8.0 Tris-HCL buffer, 0.5 ml of 100 mM L-asparaginase substrate, 0.5 milliliters of each crude enzyme, and distilled water. The reaction mixture was then incubated for 30 minutes in a water bath at 37°C. After the reaction had been incubated, point five milliliters of Trichloroacetic acid (TCA) was added to stop it. To determine nitrogen present in a sample8, 0.1 ml of the reaction mixture was added to 3.7 ml of distilled water in a separate test tube, followed by 0.2 ml of Nessler reagent. The absorbance was measured at 450 nm, and the unit of activity was calculated by comparing it to a standard curve.

#### Characterization of asparaginase

#### Effect of pH

The best pH for asparaginase activity was discovered by measuring the enzyme's activity at various pH levels. The pH stability of the enzyme was assessed by incubating it in a pH range of 5.0–10.0 for 24 hours at 4°C, and the residual activity was determined under standard assay conditions.

#### Effect of Temperature

Asparaginase activity was measured at several temperatures to determine the optimum temperature. Heat stability was determined by incubating the enzyme alone for 1 hour at various temperatures. The enzyme solution was brought to room temperature following heat treatment, and the enduring activity was measured after adding the substrates.

#### **Results**

Table 1 was shown the findings of asparaginase purification processes from Iraqi *Phaseoulus vulgaris*. Figure (1) depicted the DEAE-Sepharose column chromatography results, exhibiting three protein peaks. A Sephacryl S-200 column was used to apply peak one, which had the highest asparaginase activity. This is the last stage in the purification process and results in a single peak of the pure enzyme, as shown in Figure (2). With a specific activity of 143 units/mg protein, L-asparaginase was purified 2.7 times. After passing through the Sephacryl S-200 column, the asparaginase was found to be pure. The optimum pH for asparaginase activity was 8, as in figure (3).

At alkaline pH, the asparaginase enzyme was found to be stable. When enhancement for 24 hours as in a frame, pH fluctuated between 7.5 and 9.5 since it used 90% of its activity (4). However, the pH optimum of L-asparaginases from a few plants was shown to be between 8.0 and 8.510. Almost L- asparaginases from bacteria have alkaline pH optimum values of 8.0 to 10<sup>11</sup>.

The temperature has a significant impact on enzyme activity. The optimal temperature for asparaginase activity varies depending on the enzyme source. The optimum temperature for asparaginase was found to be 40 degrees Celsius, which was consistent with previous findings.

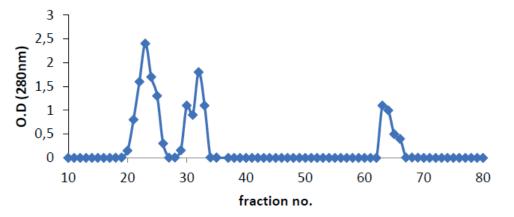
The enzyme activity was deliberate at several temperatures. To determine the optimum temperature for asparaginase stability, the enzyme was incubated at different temperature values, then the remaining activity was determined after assaying enzyme activity. As shown in figure (6), asparaginase was more stable at temperatures between 20 and 40oC because the enzyme gained maximal residual activity at these temperatures (100 percent). When the enzyme was incubated at temperatures higher than the optimum temperature for stability (35oC), the residual activity dropped, and around 45oC, asparaginase began to lose its activity.

## Discussion

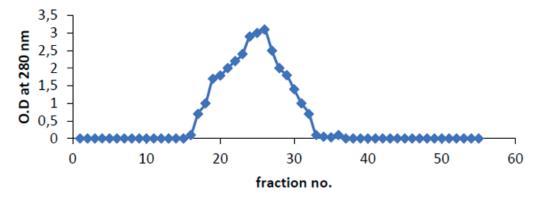
Any deviation in pH from the optimum induces ionization of R-groups of amino acids, which reduces enzyme activity; between pH 7.0 and 9.5, overmuch half of the enzyme's activity level was preserved. Even though the most significant activity at physiological pH is one of the requirements for anticancer activity in such enzymes, the revised enzyme would be feasible because 80 percent of the enzyme activity was retained at pH 8. Enzymes, like other proteins, are only stable across a narrow pH range. Changes in the charges on ionizable residues outside of this range lead to alterations to the protein's tertiary structure, which finally leads to denaturation<sup>9</sup>. At alkaline pH (pH 7.5–9.5), asparaginase appeared firm, as it employed 90%

| Step                                 | Total protein<br>(mg) | Total activity<br>(unit) | Specific activity<br>(unit/mg pro- | Fold purifica-<br>tion | Recovery % |
|--------------------------------------|-----------------------|--------------------------|------------------------------------|------------------------|------------|
| Plant crude ex-<br>tract             | 11                    | 805                      | 73                                 | 1                      | 100        |
| Dialysis                             | 5                     | 325                      | 65                                 | 0.89                   | 40         |
| DEAE-Sepharose                       | 1.2                   | 64                       | 53                                 | 0.8                    | 8          |
| Gel filtration on<br>Sephacryl S-200 | 0.3                   | 43                       | 143                                | 2.7                    | 5          |

 Table 1. Synopsis Procedures for purifying asparaginase from Iraqi green beans.



**Figure 1.** Chromatography of asparaginase from green beans on DEAE-Sepharose ( $16 \times 1.5$  cm) already to bring to balance with 20mM Tris-HCl, pH 8.0 at a flow rate of 1ml/min. and 3mL fractions.



**Figure 2.** Size exclusion chromatography of asparaginase from anion exchanger -Sepharose fraction on Sephacryl S-200 column 2.6 x 83 cm (equilibrated with tris -HCl at flow rate 0.5/minute and 3.5 ml fraction).

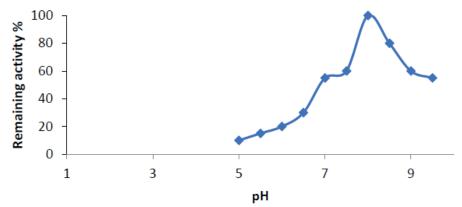
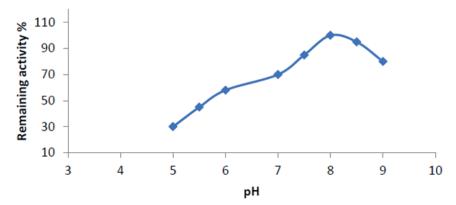
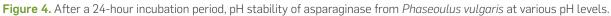


Figure 3. Optimum pH of the enzyme from green bean. Asparaginase activity was measured at different pH.





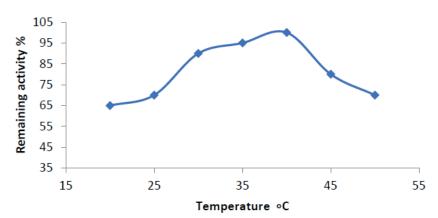


Figure 5. Optimum temperatures for purified asparaginase produced from Phaseolus vulgaris.

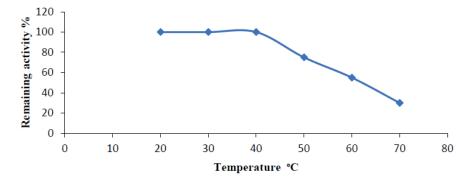


Figure 6. Stability temperature of asparaginase produced from Phaseolus vulgaris.

of its activity after being incubated for 24 hours, as shown in figure 4. However, the optimum pH of L-asparaginases from various plants ranged from eight to eight-point five<sup>10</sup>. Alkaline pH optima were found in nearly all L-asparaginases from bacteria, ranging from 8.0 to 10<sup>11</sup>. According to Ali<sup>12</sup>, the optimum temperature for the asparaginase enzyme produced by Vigna unguiculata was 40°C, and the optimum temperature for such enzyme produced by pseudomonas aeruginosa was similarly 40°C, as mentioned by (13). Denaturation of the enzyme occurs when the protein's three-dimensional structure is destroyed, resulting in the creation of random polypeptide chains as the temperature rises. This produces a change in the active site, rendering the enzyme inactive at high temperatures<sup>14</sup>. Pseudomonas acidovorans glutaminase decreased roughly 50% of its activity after being incubated at 50°C for 10 minutes, according to Davidson<sup>15</sup>. In contrast, Koibuhi<sup>16</sup> and Daiwa<sup>17</sup> found that glutaminase from *B. subtilis* kept all its activity after one hour of incubation at 50°C. In another bacterial enzyme, the Optimum optimal temperature for activity was 37°C, but the stability of chondroitinase was maintained 100% at 20-40°C for 30 min 18. As the temperature rises, enzyme activity may decrease, resulting in denaturation, which entails the loss of the protein's three-dimensional structure and the formation of random amino acids.

# Conclusions

The enzyme L-asparaginase was isolated from Iraqi green beans so that any potential adverse effects could be ruled out if anticancer therapy is investigated. Asparaginase showed good consistency across various physiological environments, including pH and temperature. We should investigate asparaginase as cancer or leukemia treatment in the future.

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#### **Conflicts of Interest**

None.

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